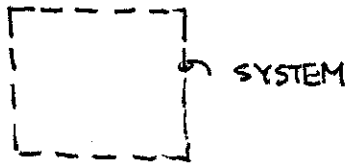
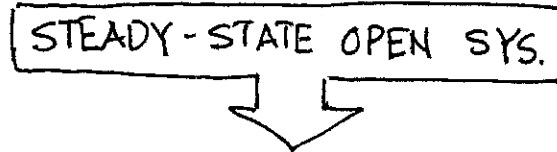
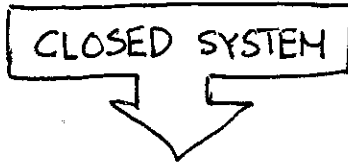


FOR A GENERAL SYSTEM →



Cons. of Mass:

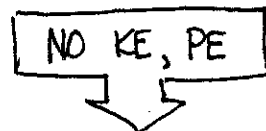
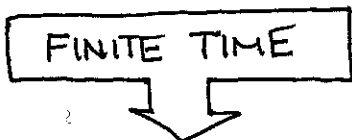
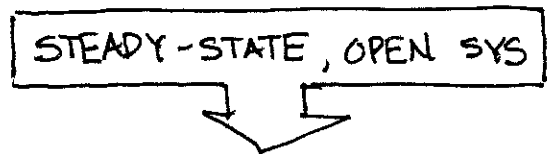
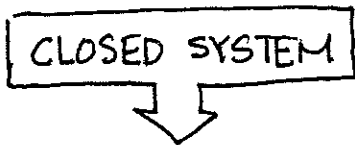
$$\frac{dm_{\text{sys}}}{dt} = \sum_{\text{IN}} \dot{m} - \sum_{\text{OUT}} \dot{m}$$



Cons. of Energy:

$$\frac{dE_{\text{sys}}}{dt} = \dot{Q}_{\text{IN}} + \dot{W}_{\text{IN}} + \sum_{\text{IN}} \dot{m} \left( h + \frac{V^2}{2} + gz \right) - \sum_{\text{OUT}} \dot{m} \left( h + \frac{V^2}{2} + gz \right)$$

\* NO KE or PE ⇒



# PROPERTY STUFF

1. INCOMPRESSIBLE SUBSTANCE W/ CONST. C

$$\rho = \frac{1}{v} = \text{CONST}$$

$$u_2 - u_1 =$$

$$h_2 - h_1 = u_2 - u_1 + v(P_2 - P_1) =$$
$$=$$

$$P_2 - P_1 =$$

2. IDEAL GAS W/ CONST  $c_v, c_p$

$$Pv = NR_u T$$

$$R = \frac{R_u}{M}$$

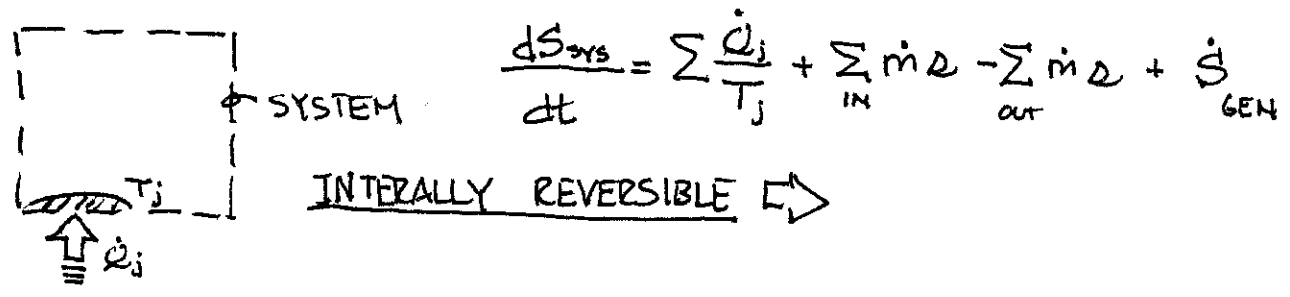
$$u_2 - u_1 =$$

$$h_2 - h_1 =$$

$$P_2 - P_1 =$$
$$=$$

# ENTROPY ACCT'ING & 2ND LAW STUFF

FOR A GENERAL SYSTEM:



CLOSED SYSTEM

OPEN, STEADY-STATE

FINITE TIME

THE 2ND LAW SAYS...

$\dot{S}_{gen}$

## OTHER USEFUL STUFF

• Boundary work for a closed system:

$$W_{1 \rightarrow 2} =$$

• Mass flow rate across a surface

$$\dot{m} =$$

• "Td $\Delta$ " Equations

$$T d\Delta =$$

$$T d\Delta =$$