

Example—Property Table Practice

Given: water, $P = 10 \text{ MPa}$, $T = 350^\circ\text{C}$

Find: v (spec. vol.)

P-TABLE: $T_{\text{SAT}} = 311^\circ\text{C} < T \Rightarrow$ SHU

② T-TABLE:

$$\text{SHU: } \frac{v - 0.01925}{0.02331 - 0.01925} = \frac{350 - 320}{360 - 320}$$

$$v = 0.022295$$

$$= \boxed{0.02230 \text{ m}^3/\text{kg}}$$

Given: water, $P = 10 \text{ MPa}$, $T = 360^\circ\text{C}$

Find: v (spec. vol.)

P-TABLE: $T_{\text{SAT}} = 311^\circ\text{C} < T \Rightarrow$ SHU

① T-TABLE: $P_{\text{SAT}} = 186.5 \text{ bar} > P \Rightarrow$ SHU

$$\boxed{v = 0.02331 \text{ m}^3/\text{kg}}$$

Given: water, $T = 360^\circ\text{C}$, $v = 0.003 \text{ m}^3/\text{kg}$

Find: P, h

③ SAT T-TABLE

$$P = 186.5 \text{ bar}$$

$$v = (1-x)v_f + xv_g = (1-x)(0.001893) + x(0.006945)$$

$$x = 0.2191$$

$$h = (1-x)h_f + xh_g$$

$$= (1-0.2191)(1760.5) + (0.2191)(2481)$$

$$= \boxed{1918 \text{ kJ/kg}}$$

Given: water, $T = 20^\circ\text{C}$, $P = 10\text{ MPa}$

Find: s (spec. entropy)

$$P_{\text{SAT}} = 2.339\text{ kPa}$$

$$P \gg P_{\text{SAT}} \Rightarrow \text{C.L.}$$

C.L. TABLES

$$\Delta = 0.2945\text{ kJ/kg}\cdot\text{K}$$

Given: water, $T = 20^\circ\text{C}$, $P = 4\text{ kPa}$

Find: s (spec. entropy)

NO CL TABLES FOR THIS RANGE!!!

COMPARE

$$v_f (T=20^\circ\text{C}) = 0.001002$$

$$P = 2.339\text{ kPa}$$

$$\Delta_f = 0.2966$$

$$v (T=20^\circ\text{C}) = 0.0009995$$

$$P = \underline{\underline{5\text{ MPa}}}$$

$$\Delta = 0.2956$$

SMALL
CHANGE

SMALL CHANGE

$$\therefore u \approx u_f(T)$$

$$\Delta \approx \Delta_f(T)$$

$$v \approx v_f(T)$$

$$h \approx h_f + v_f(P - P_{\text{SAT}})$$

$$\cancel{u \approx u_f(P)}$$