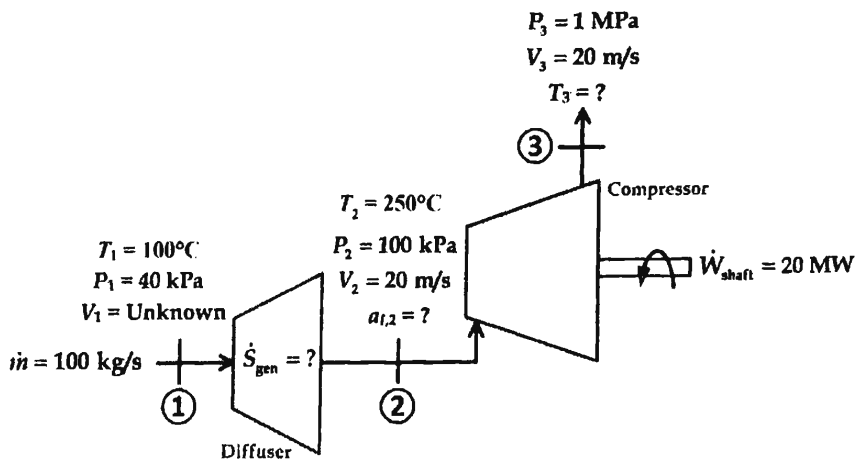


PROBLEM 2 [35 points]

Air flows through a diffuser and compressor at a rate of 100 kg/s and with the state properties as shown in the figure.

Model air as an ideal gas with variable specific heats and assume surrounding conditions of $P_0 = 101.325 \text{ kPa}$ and $T_0 = 298 \text{ K}$.



Determine:

- 10 (a) the rate of entropy generation in the diffuser, in kW/K,
- 12 (b) the temperature of the air at the exit of the compressor assuming that the compressor requires a power input of 20 MW, in K, and
- 17 (c) the specific flow exergy into the compressor, in kJ/kg.

(a)

$$\frac{d(S_{sys})}{dt} = \sum \frac{\dot{Q}_i}{T_{b,i}} + \dot{m} \Delta_1 - \dot{m} \Delta_2 + \dot{S}_{gen}$$

Steady

$$\dot{S}_{gen} = \dot{m} (\Delta_2 - \Delta_1)$$

$$\dot{S}_{gen} = (\dot{m}) \left(\Delta_{T_2}^0 - \Delta_{T_1}^0 - R \ln \frac{P_2}{P_1} \right)$$

$$= \left(100 \frac{\text{kg}}{\text{s}} \right) \left[\left(2.26588 - 1.92112 \right) \frac{\text{kJ}}{\text{kg} \cdot \text{K}} - 0.287 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} \ln \left(\frac{100 \text{ kPa}}{40 \text{ kPa}} \right) \right]$$

$$= 8.18 \text{ kW/K} \leftarrow \text{ANS}$$

(b)

COE

$$\frac{d(E_{sys})}{dt} = \dot{Q}_{in} + \dot{W}_{in} + \dot{m} (h_2 + \dots) - \dot{m} (h_3 + \dots)$$

Steady

$$h_3 = h_2 + \frac{\dot{W}_{in}}{\dot{m}} = 526.74 \frac{\text{kJ}}{\text{kg}} + \frac{20000 \text{ kW}}{100 \frac{\text{kg}}{\text{s}}} \left(\frac{\text{kJ/s}}{\text{kJ}} \right)$$

$$h_3 = 726.74 \text{ kJ/kg}$$

$$h_2 = h(T_2) = 526.74 \frac{\text{kJ}}{\text{kg}} \quad h_3 = 726.74 \text{ kJ/kg}$$

$$T_3 = T(h_3) = 712.5 \text{ K} \leftarrow \text{ANS}$$

$$(c) a_{f,2} = h_2 - h_0 - T_0(\Delta_2 - \Delta_0) + ke$$

$$= h_2 - h_0 - T_0 \left[\Delta_{T_2}^\circ - \Delta_{T_0}^\circ - R \ln \left(\frac{P_2}{P_0} \right) \right] + \frac{V_2^2}{2}$$

$$h(T_2) = 526.7 \text{ kJ/kg}$$

$$h(T_0) = 298.18 \text{ "}$$

$$\Delta^\circ(T_2) = 2.26588 \text{ kJ/kg}\cdot\text{K}$$

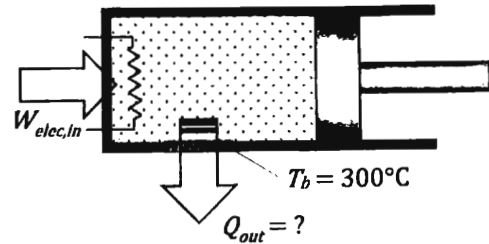
$$\Delta^\circ(T_0) = 1.69533 \text{ "}$$

$$a_{f,2} = (526.7 - 298.18) \frac{\text{kJ}}{\text{kg}} - (298 \text{ K}) \left[(2.26588 - 1.69533) \frac{\text{kJ}}{\text{kg}\cdot\text{K}} \right. \\ \left. - 0.287 \frac{\text{kJ}}{\text{kg}\cdot\text{K}} \ln \left(\frac{100 \text{ kPa}}{101.325 \text{ kPa}} \right) \right] + \frac{20^2 \text{ m}^2}{8^2} \left\langle \frac{\text{kJ/kg}}{1000 \text{ m}^2/\text{s}^2} \right\rangle$$

$$= 59.6 \frac{\text{kJ}}{\text{kg}} - 0.2 \frac{\text{kJ}}{\text{kg}} = 59.4 \frac{\text{kJ}}{\text{kg}} \quad \leftarrow \text{ANS}$$

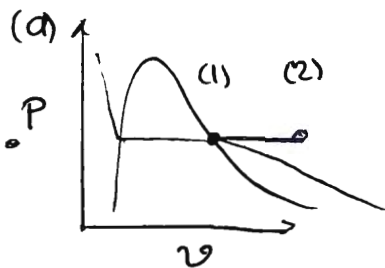
Problem 3 [45 points]

A mass of $m = 2.0$ kg of water inside a piston-cylinder device is initially as a saturated vapor at a pressure of $P_1 = 400$ kPa. A resistance heater, placed inside the cylinder, is turned on and delivers 4 kW of electrical power for 10 minutes as the system expands under constant pressure until its volume reaches $V_2 = 2.0$ m³.



Use $T_0 = 25^\circ\text{C}$ and $P_0 = 101.325$ kPa.

- 5 (a) Sketch the process path from the initial to the final state on a $P-v$ diagram. Clearly indicate the relative position of the two-phase dome.
- 5 (b) Determine the final temperature of water, in $^\circ\text{C}$.
- 20 (c) Determine the heat loss to the surroundings, in kJ.
- 15 (d) Assume all heat loss occurs at an average boundary temperature of 300°C , determine the exergy destruction for the entire process, in kJ.

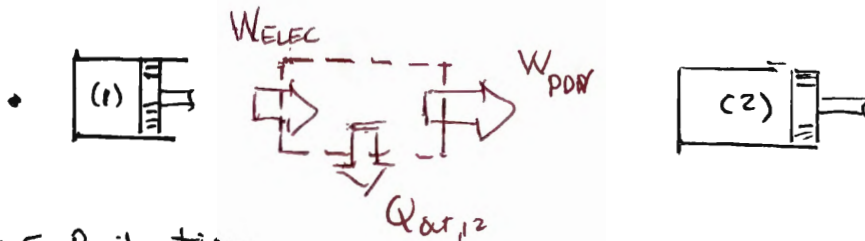


$$(b) P_2 = P_1 = 400 \text{ kPa}$$

$$v_2 = V_2/m = 2.0 \text{ m}^3 / 2.0 \text{ kg} = 1.0 \text{ m}^3/\text{kg}$$

$$\bullet T_2 = T(P_2, v_2) = 59.5.2^\circ\text{C} \leftarrow \text{ANS}$$

(c)



CoE, finite time,
closed system, no KE/PE

$$\bullet U_2 - U_1 = -Q_{out,12} + \bar{W}_{NET,IN} = -Q_{out,12} + W_{ELEC,IN} - W_{PDV}$$

$$m(u_2 - u_1) = -Q_{out,12} + \bar{W}_{ELEC,IN} - W_{PDV}$$

$$Q_{out,12} = \bar{W}_{ELEC,IN} - \bar{W}_{PDV} - m(u_2 - u_1)$$

$$\bullet \bar{W}_{PDV} = \int_{v_1}^{v_2} P dv = P(v_2 - v_1) = mP_1(v_2 - v_1)$$

$$\bullet v_1 = v(P_1, x_1=1) = 0.4627 \text{ m}^3/\text{kg}$$

$$W_{POW} = (20 \text{ kg})(400 \text{ kPa})(1.0 - 0.4627) \text{ m}^3/\text{kg} \left\langle \frac{\text{kJ}}{\text{kPa} \cdot \text{m}^3} \right\rangle$$

$$= \underline{430 \text{ kJ}}$$

$$\bullet W_{ELEC} = \dot{W}_{ELEC} \cdot \Delta t = (4.0 \text{ kW})(10 \text{ minutes}) \left\langle \frac{60 \text{ s}}{\text{min}} \right\rangle \left\langle \frac{\text{kJ}}{\text{kW} \cdot \text{s}} \right\rangle$$

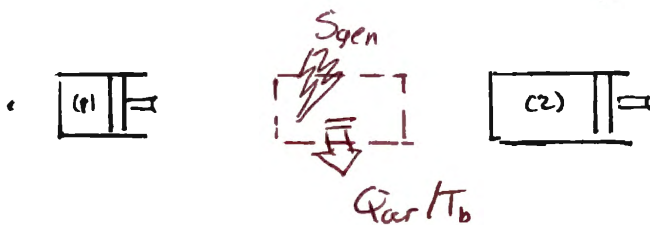
$$= \underline{2400 \text{ kJ}}$$

$$\bullet \begin{cases} u_1 = u(P_1, x_1) = 2553.4 \text{ kJ/kg} \\ u_2 = u(P_2, u_2) = 3292.1 \text{ "} \end{cases}$$

$$\bullet Q_{out,12} = 2400 \text{ kJ} - \dots \text{ kJ} - 20 \text{ kg}(3292.1 - 2553.4) \text{ kJ/kg}$$

$$= 491 \text{ kJ} \leftarrow \text{ANS}$$

(d)



AoS: closed, finite

$$(S_2 - S_1)_{sys} = \sum \frac{Q_{in,i}}{T_{b,i}} + S_{gen}$$

$$\bullet S_{gen} = m(\Delta s_2 - \Delta s_1) + Q_{out}/T_0$$

$$= (20 \text{ kg})(8.444 - 6.896) \frac{\text{kJ}}{\text{kg} \cdot \text{K}}$$

$$+ 464 \text{ kJ}/(300 + 273) \text{ K} = 3.91 \text{ kJ/K}$$

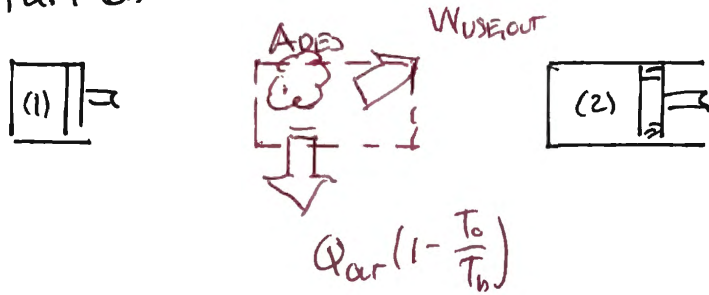
$$\bullet \begin{cases} \Delta s_1 = \Delta s(P_1, x_1) = 6.896 \text{ kJ/kg} \\ \Delta s_2 = \Delta s(P_2, u_2) = 8.444 \text{ kJ/kg} \end{cases}$$

$$\bullet A_{DES} = T_0 \cdot S_{gen}$$

$$= (25 + 273) \text{ K}(3.91 \text{ kJ/K})$$

$$\bullet = 1164 \text{ kJ} \leftarrow \text{ANS}$$

Part (d) via AoA:



AoA, finite, closed, no KE/PE

$$(A_2 - A_1)_{sys} = \sum \underline{Q}_{in} (1 - T_0/T_b) - \bar{W}_{USE,OUT} - A_{DES}$$

$$A_{DES} = (A_1 - A_2)_{sys} - Q_{out} (1 - T_0/T_b) + \bar{W}_{IN,ELEC} - [\bar{W}_{OUT,PDV} - P_0 (V_2 - V_1)]$$

$$= m(a_2 - a_1) - \dots$$

$$= m(u_2 - u_1 + P_0(v_2 - v_1) - T_0(s_2 - s_1)) - Q_{out} (1 - \frac{T_0}{T_b}) + \bar{W}_{IN,ELEC} - \bar{W}_{OUT,PDV} + P_0 (V_2 - V_1)$$

$$= (20 \text{ kg}) \left[(2553.4 - 3292.1) \frac{\text{kJ}}{\text{kg}} + 101.325 \text{ kPa} (0.4267 - 1.0) \frac{\text{m}^3}{\text{kg}} - (298 \text{ K}) (6.898 - 8.444) \frac{\text{kJ}}{\text{kg} \cdot \text{K}} \right]$$

$$- 464 \text{ kJ} \left(1 - \frac{298 \text{ K}}{1300 + 273 \text{ K}} \right) + 2400 \text{ kJ} - 459.5 \text{ kJ}$$

$$+ 101.325 \text{ kPa} (1.0 - 0.4267) \text{ m}^3/\text{kg}$$

$$= 1164 \text{ kJ} \leftarrow$$

ANS