

Section: 01 (Lui, 9th hour) 04 (Mech, 10th hour)
 02 (Lui, 10th hour) 05 (Adams, 9th hour)
 03 (Mech, 9th hour) 06 (Adams, 10th hour)

Name

CM

Exam 1

Oct 26, 2015

Problem 1	_____ / 15
Problem 2	_____ / 40
Problem 3	_____ / 45
Total	_____ / 100

Show all work for full credit.

Open property tables. **Do not interpolate;** use closest values.

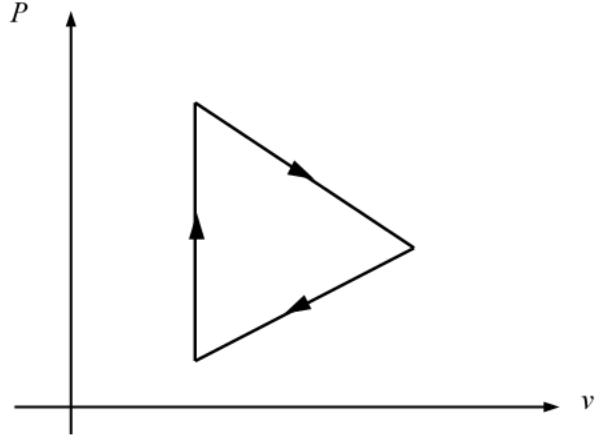
Computer use for computational purposes and/or music.

Two 8 ½ × 11" equation sheets – one side, hand-written, no worked examples or homework.

CM: _____

Problem 1 [15 pts]

- (a) [6 pts] Consider a thermodynamic cycle of a closed system as represented by the following P - v diagram. Does it describe a power cycle or a refrigeration cycle? Briefly explain your answer.



- (b) [9 pts] Circle the correct answer(s) in the following questions:

- a. [3 pts] The thermal efficiency of a power cycle is a second-law efficiency.

True | False

- b. [2 pts] Inter-cooling between successive stages of compression has the following effect on the net power output:

Increase | Decrease | No change

- c. [2 pts] An increase in the compression ratio of a reciprocating internal combustion engine has the following effect on the overall thermal efficiency of the cycle

Increase | Decrease | No change

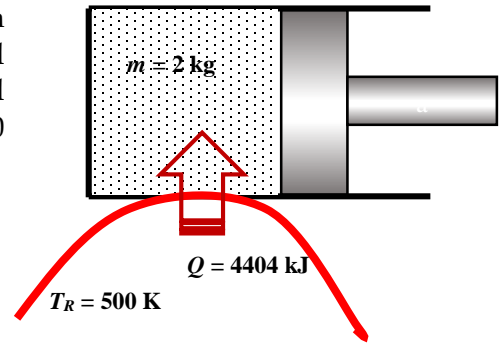
- d. [2 pts] Under steady-state operation, a fluid at a given location experiences no change in the following thermodynamic variable(s). Circle all correct answer(s).

Pressure | Enthalpy | Entropy

Problem 2 [40 pts]

A piston cylinder contains a mass $m = 2$ kg of saturated liquid steam at a pressure of $P_1 = 2$ bar. Heat transfer in the amount of $Q = 4404$ kJ is added to the steam and it expands at constant pressure until it is a saturated vapor. The heat transfer comes from a reservoir at temperature $T_R = 500$ K. The surroundings are at $T_0 = 300$ K and $P_0 = 100$ kPa.

- (a) [15 pts] Find the useful work into or out of the steam indicate its direction.
- (b) [10 pts] Find the change in exergy of the steam during the process.
- (c) [15 pts] **Using an accounting of exergy approach** (i.e., not entropy accounting), find the total exergy destroyed in this process.



CM: _____

CM: _____