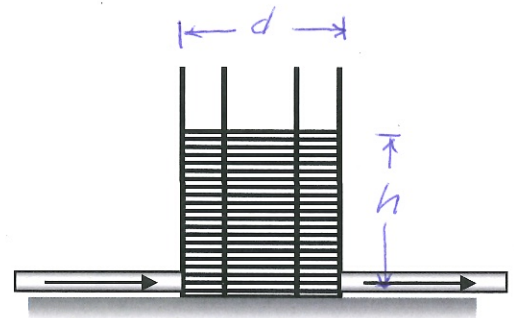


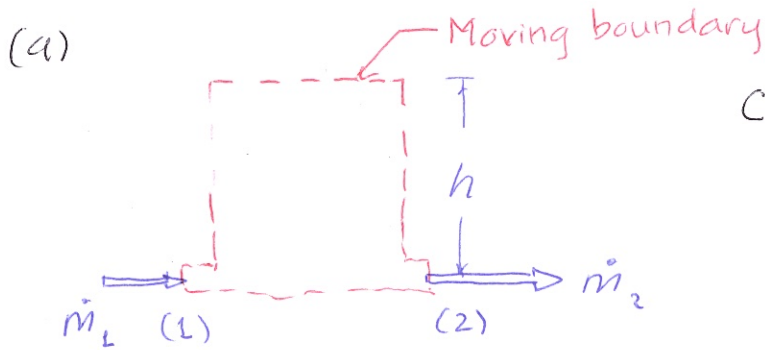
Example

On the east coast, natural gas is frequently stored in large flexible tanks that can expand and contract while maintaining a relatively constant pressure. The walls of the tank are made of a flexible accordion membrane, and the roof and floor of the tank are rigid. As the tank fills up the roof of the tank "floats" up. The vertical tank has diameter  $d = 50$  ft and a height  $h$  which change from 5 ft to 30 ft. Under normal operating conditions, the temperature and pressure inside the tank is uniform and constant. The density of the natural gas in the tank under these conditions is  $\rho = 0.12$  lbm/ft<sup>3</sup>.

- (a) Under one set of operating conditions,  $h = 10$  ft and natural gas is flowing into the tank at 470 lbm/min and flowing out of the tank at 235 lbm/min, and natural gas has a density  $\rho = 0.12$  lbm/ft<sup>3</sup>.
  - Is the roof of the tank moving up or down? How fast?
  - How far will it move in 5 minutes?
- (b) On another day, the tank operators observe a drop in the roof height  $h$  even though all tank valves are shut and they are worried there could be a leak. Or it might just be caused by the change in temperature due to the cold snap. Initially, the roof height  $h_1 = 10$  ft when the air (and natural gas) temperature  $T_1 = 40^\circ\text{F}$  (500°R). And after the cold snap, the temperature  $T_2 = 10^\circ\text{F}$  (470°R).



If the tank pressure is uniform and constant and natural gas (methane) can be modeled as an ideal gas with  $R_{\text{methane}} = 96.3$  ft-lbf/lbm-°R, determine the new roof height after the cold snap.



Cons. of mass  $\rightarrow$

$$\frac{d}{dt}(M_{\text{sys}}) = \sum \dot{m}_{\text{in}} - \sum \dot{m}_{\text{out}}$$

$$M_{\text{sys}} = \rho V_{\text{sys}} = \rho \frac{\pi d^2}{4} h$$

$$\therefore \frac{d}{dt} \left( \rho \frac{\pi d^2}{4} h \right) = \dot{m}_1 - \dot{m}_2$$

$$\rho \frac{\pi d^2}{4} \frac{dh}{dt} = \dot{m}_1 - \dot{m}_2$$

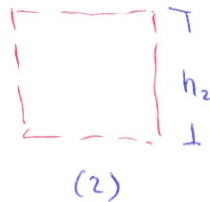
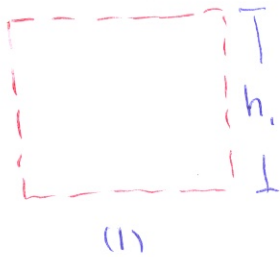
$$\frac{dh}{dt} = \frac{4(\dot{m}_1 - \dot{m}_2)}{\rho \pi d^2} = \frac{4 \left( 470 \frac{\text{lbm}}{\text{min}} - 235 \frac{\text{lbm}}{\text{min}} \right)}{0.12 \frac{\text{lbm}}{\text{ft}^3} \cdot \pi \cdot 50^2 \text{ft}^2} = 0.997 \frac{\text{ft}}{\text{min}}$$

It is moving  $\sim 1 \text{ ft/min}$  up.

In 5 mins (since  $dh/dt = \text{const}$ )

$$\text{dist} = \frac{dh}{dt} \cdot \text{time} = (1 \text{ ft/min}) \cdot (5 \text{ min}) = \boxed{5.0 \text{ ft}}$$

(b)



Cons. of mass

$$\frac{d}{dt}(m_{\text{sys}}) = \sum \dot{m}_{\text{in}} - \sum \dot{m}_{\text{out}}$$

closed.

$$m_{\text{sys}} = \rho V$$

$$= \rho \cdot \frac{\pi d^2}{4} h$$

$$= \left(\frac{P}{RT}\right) \frac{\pi d^2}{4} h \quad (1)$$

$$\frac{d}{dt}(m_{\text{sys}}) = 0$$

$$\int_{m_1}^{m_2} dm_{\text{sys}} = \int_{t_1}^{t_2} 0 \cdot dt$$

$$m_{\text{sys},2} - m_{\text{sys},1} = 0$$

$$m_{\text{sys},2} = m_{\text{sys},1}$$

(For an ideal gas

$$P = \rho RT)$$

Using (1):

$$\frac{\rho_2}{R_{\text{METHANE}} T_2} \frac{\pi d^2}{4} h_2 = \frac{\rho_1}{R_{\text{METH}} T_1} \frac{\pi d^2}{4} h_1$$

$$h_2 = \frac{T_2}{T_1} h_1 = \frac{470^\circ\text{R}}{500^\circ\text{R}} (10 \text{ ft}) = \boxed{9.4 \text{ ft}}$$