

Name: \_\_\_\_\_ Campus Mail Box \_\_\_\_\_

Problem 1 ( 29 ) \_\_\_\_\_

Problem 2 ( 29 ) \_\_\_\_\_

Problem 3 ( 42 ) \_\_\_\_\_  
\_\_\_\_\_

TOTAL ( 100 ) \_\_\_\_\_

**General Comments**

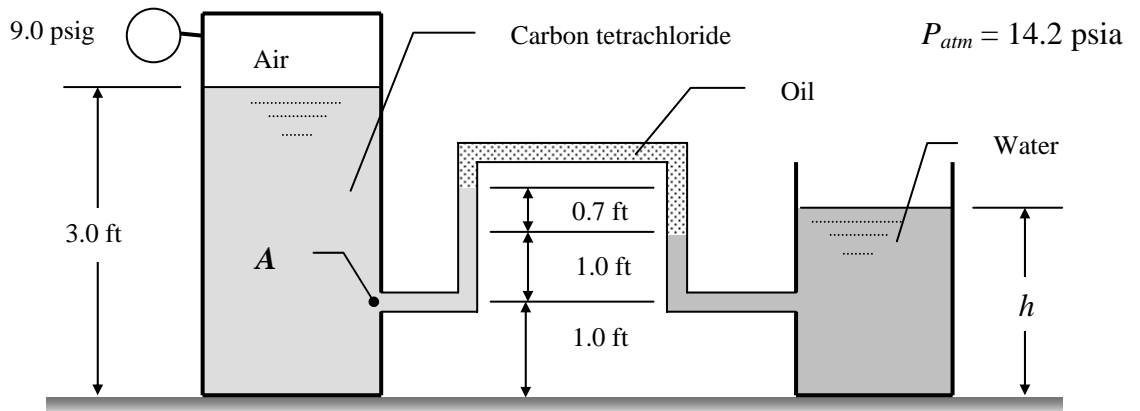
- (1) Anytime you apply conservation or accounting principles in solving a problem, sketch and clearly identify the system you have selected. In addition, clearly indicate how your assumptions or given information simplifies the general equations. Numbered-symbols, e.g.  $P_1$ , in equations must make sense for the problem in question, i.e. they must match you figure and communicate information accurately.
- (2) Closed book/notes; however, you may use any of the following:
  - ... yellow equation pages provided by instructor
  - ... unit conversion page
  - ... material in Appendices and end flaps of Wark/Richards text
  - ... your equation page (single side of 8-1/2 x 11 sheet of paper)
- (3) For maximum credit,
  - ... solve problems symbolically first showing logic and reasoning for solution,
  - ... substitute numbers into the equations clearly showing any required unit conversion factors
  - ... then and only then crunch numbers on your calculator.If I only have to punch your numbers into a calculator to get a correct answer (including units) you will receive full credit. **Don't make me guess what you are doing and why you chose to do this.**
- (4) Watch the time and feel free to remove the staple and take the test apart so that you don't have to keep flipping pages around.

**PLEASE REMOVE THE STAPLE AND USE EXTRA PAPER *INSTEAD OF* WRITING ON THE BACK OF PAGES AND THEN HAVING TO FLIP BACK AND FORTH TO FINISH THE PROBLEM. I'LL GLADLY RESTAPLE YOUR TEST!**

**Problem 1** ( 29 points )

An inverted U-tube manometer containing oil (SG = 0.8) is located between two reservoirs as shown in the figure. The reservoir on the left, which contains carbon tetrachloride ( $\gamma_{\text{carbon tetrachloride}} = 99.5 \text{ lbf/ft}^3$ ), is closed and pressurized with air. The pressure gage attached to the tank indicates that the air pressure in the tank is 9.0 psig. The reservoir on the right contains water ( $\gamma_{\text{water}} = 62.4 \text{ lbf/ft}^3$ ) and is open to the atmosphere. Assume atmospheric pressure is measured to be 14.2 psia, and there is no fluid flowing between the reservoirs.

- (a) Determine the pressure at point A, in psia .
- (b) Determine the depth of the water,  $h$ , in the right reservoir.

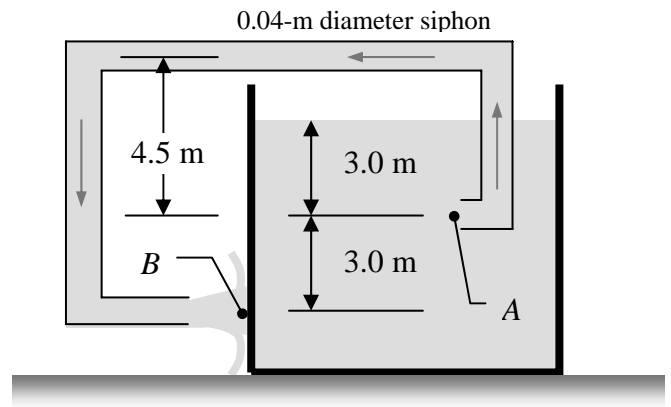


**Problem 2** ( 29 points )

Water is siphoned from a tank as shown in the figure at right. Neglect losses in the siphon.

Water Data:  $\rho = 999 \text{ kg/m}^3$ ;  $\gamma = 9.80 \text{ kN/m}^3$

- (a) Determine the volumetric flow rate through the siphon.
- (b) Determine the pressure at point *A*, the inlet to the siphon. Report your answer in either kPa (abs) or kPa (gage).
- (c) Determine the pressure at point *B*, a stagnation point. Report your answer in either kPa (abs) or kPa (gage).



**Problem 3** ( 42 points )

An air turbine attached to a compressed air tank is used to start an emergency diesel engine-generator.

The compressed air tank has a regulator attached so that the inlet pressure to the air turbine is maintained at 300 kPa and the inlet temperature is measured to be 27°C. The air leaves the turbine at 100 kPa and -33°C. Under steady-state, adiabatic conditions the turbine delivers 15 kilowatts of shaft power to the diesel engine.

Assume that air can be modeled as an ideal gas. (*If you desire* you may treat its specific heat as a constant at the value of the inlet temperature, i.e.  $c_p = 1.005 \text{ kJ}/(\text{kg}\cdot\text{K})$ ). Changes in kinetic and gravitational potential energy are negligible.

- (a) Determine the mass flow rate of air through the turbine, in kg/s.
- (b) Determine the adiabatic efficiency of the air turbine.
- (c) Sketch the process for the air flowing through the turbine on a  $T$ - $s$  diagram.

